

Towards NICE-OHMS for the detection of peroxy radicals



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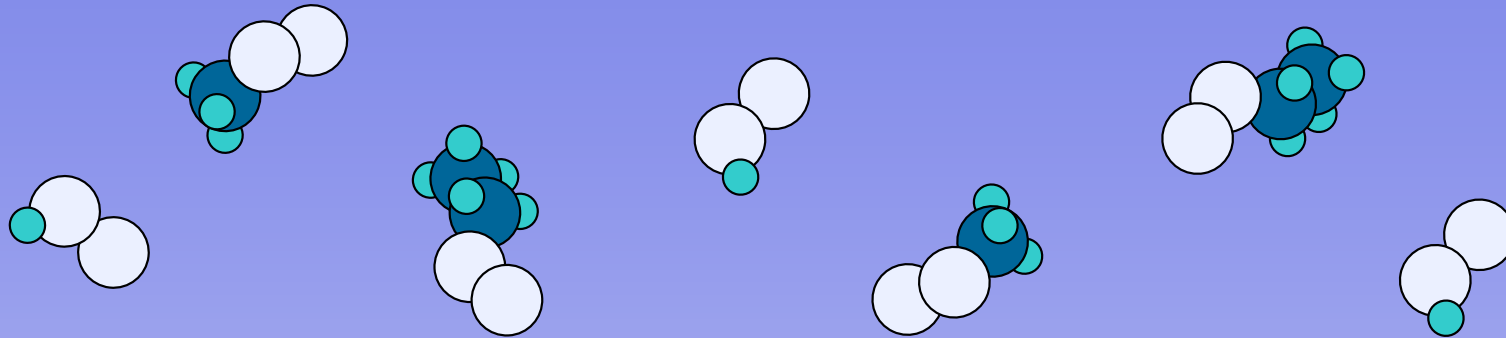
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University of Oxford



Outline

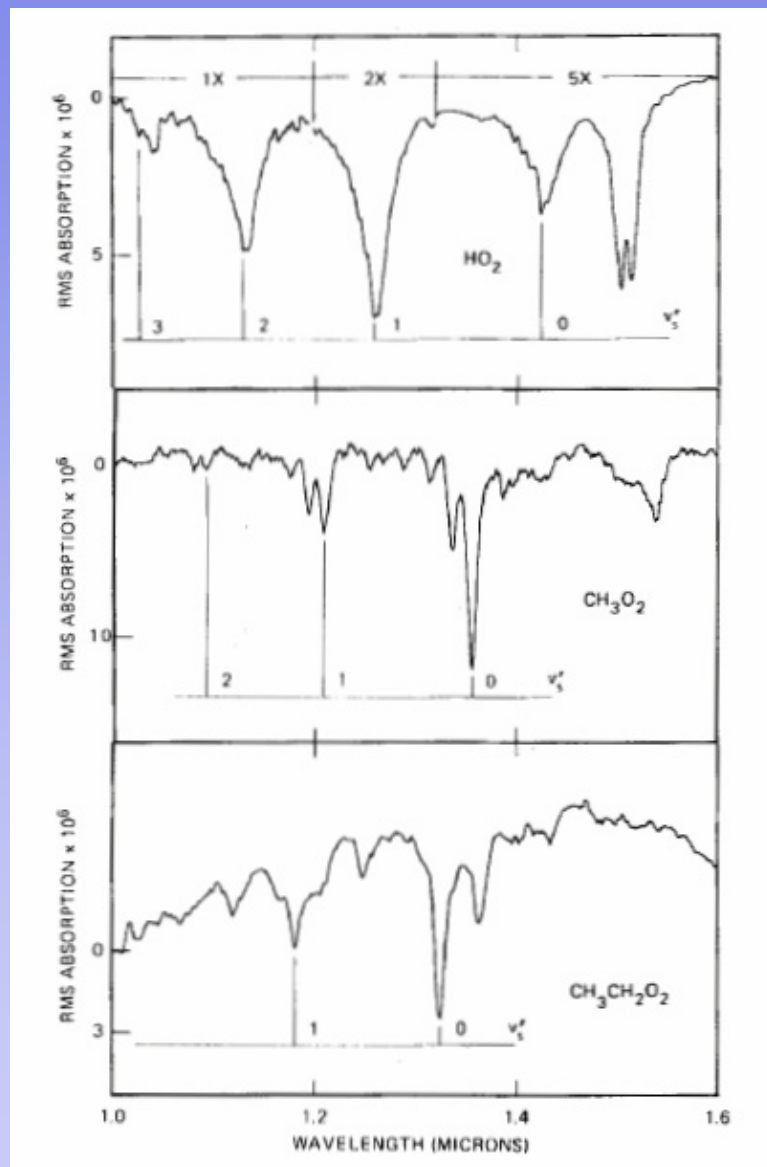
- Peroxy radicals
 - Requirements for direct detection
- NICE-OHMS and the Oxford setup
 - Steps towards NICE-OHMS
 - Initial results
 - Improving the system
- Plans and Prospects

Peroxy radicals



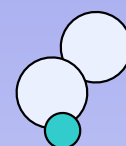
- RO₂ molecules (HO₂ , CH₃O₂ etc)
- Important intermediate in breakdown of organic pollutants
- Concentrations of 10⁷ – 10⁸ molecules cm⁻³
- Current measurements by PERCA and FAGE

Direct detection of peroxy radicals



- Distinct absorption bands in the near infra-red

- Need to detect in one of the atmospheric windows



HO_2 at ~ 1510 nm

- Atmospheric detection of HO_2 at 1510 nm requires sensitivity of order 10^{-13} to 10^{-14} cm^{-1}

Hunziker, HE and Wendt HR,
J. Chem. Phys., **64**, 8, 3488 (1976)

NICE-OHMS

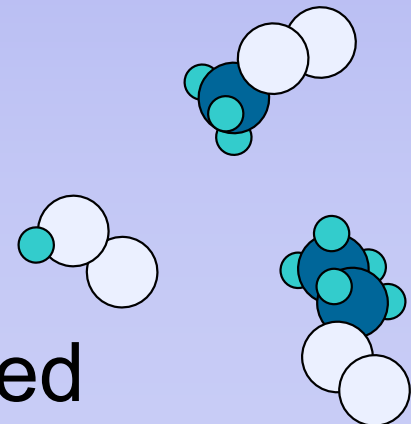
Noise-immune cavity-enhanced optical heterodyne
molecular spectroscopy

NICE-OHMS combines

- Cavity enhanced spectroscopy
 - Increase absorption path length
- Frequency modulation spectroscopy
 - Decrease noise level

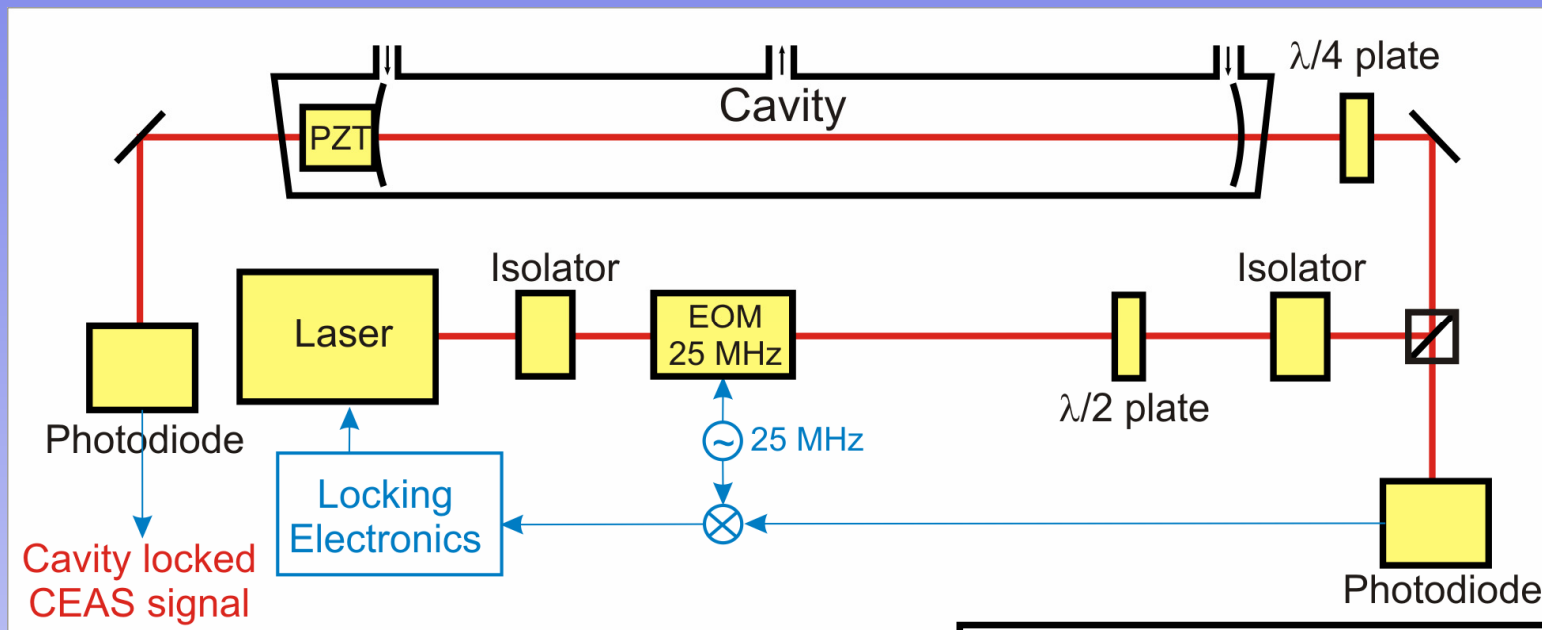
To Give

- Ultra high sensitivity
 - 10^{-14} cm^{-1} in 1s demonstrated

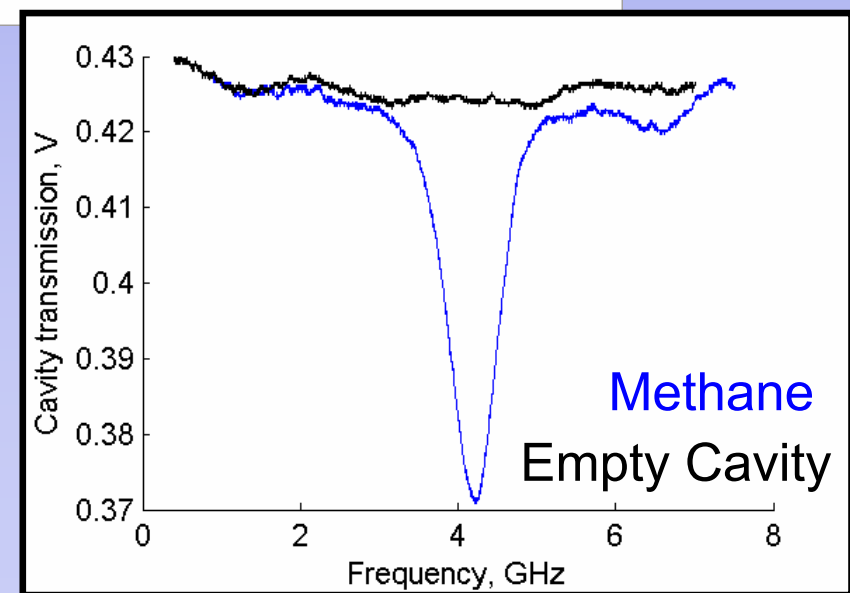


Oxford Experiment – Step 1

Cavity locked CEAS

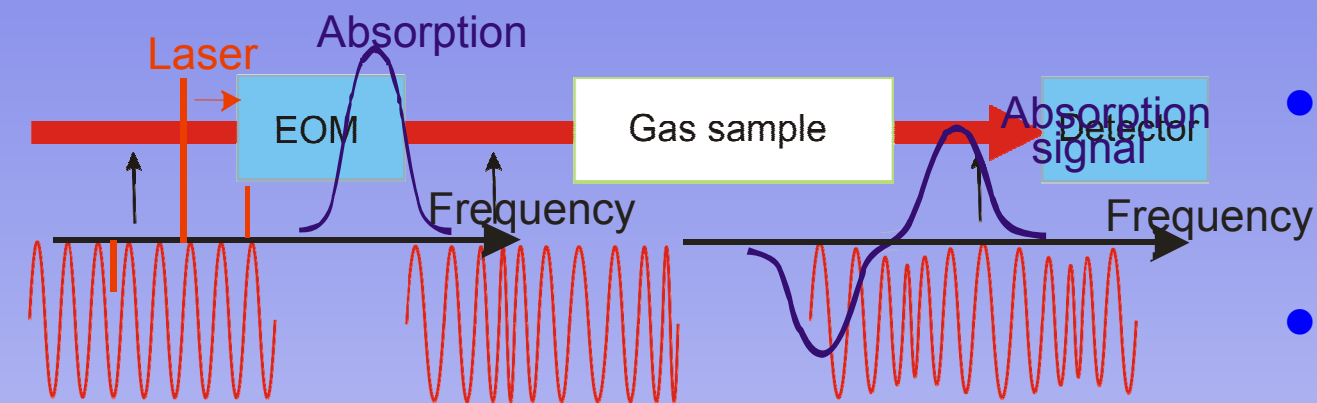


- Laser electronically locked to high finesse cavity
- Cavity transmission gives absorption profile
- Sensitivity of $8.37 \times 10^{-9} \text{ cm}^{-1}$

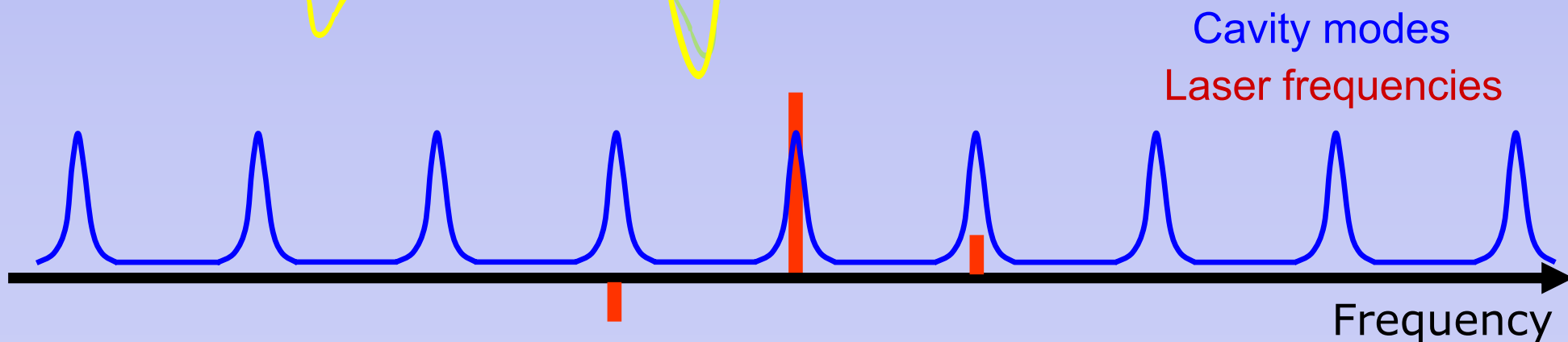
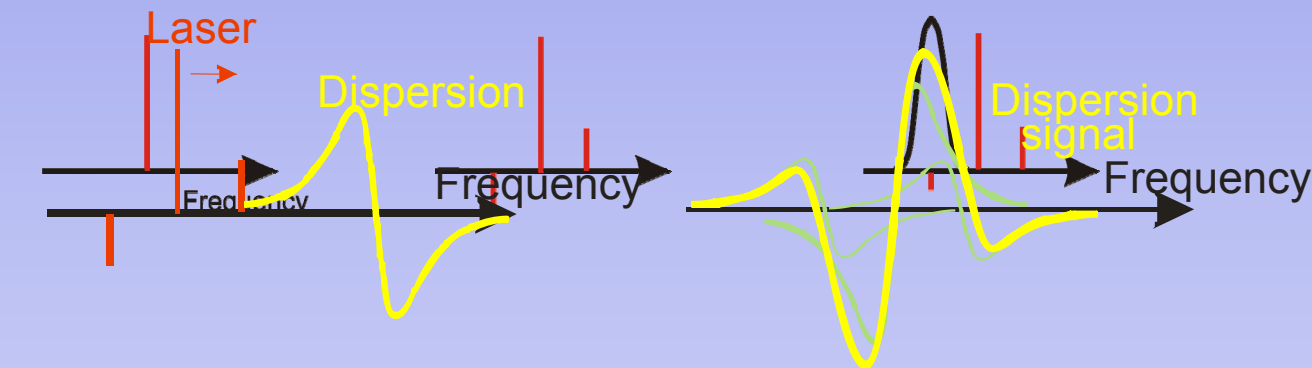


Oxford Experiment – Step 2

Cavity enhanced FMS

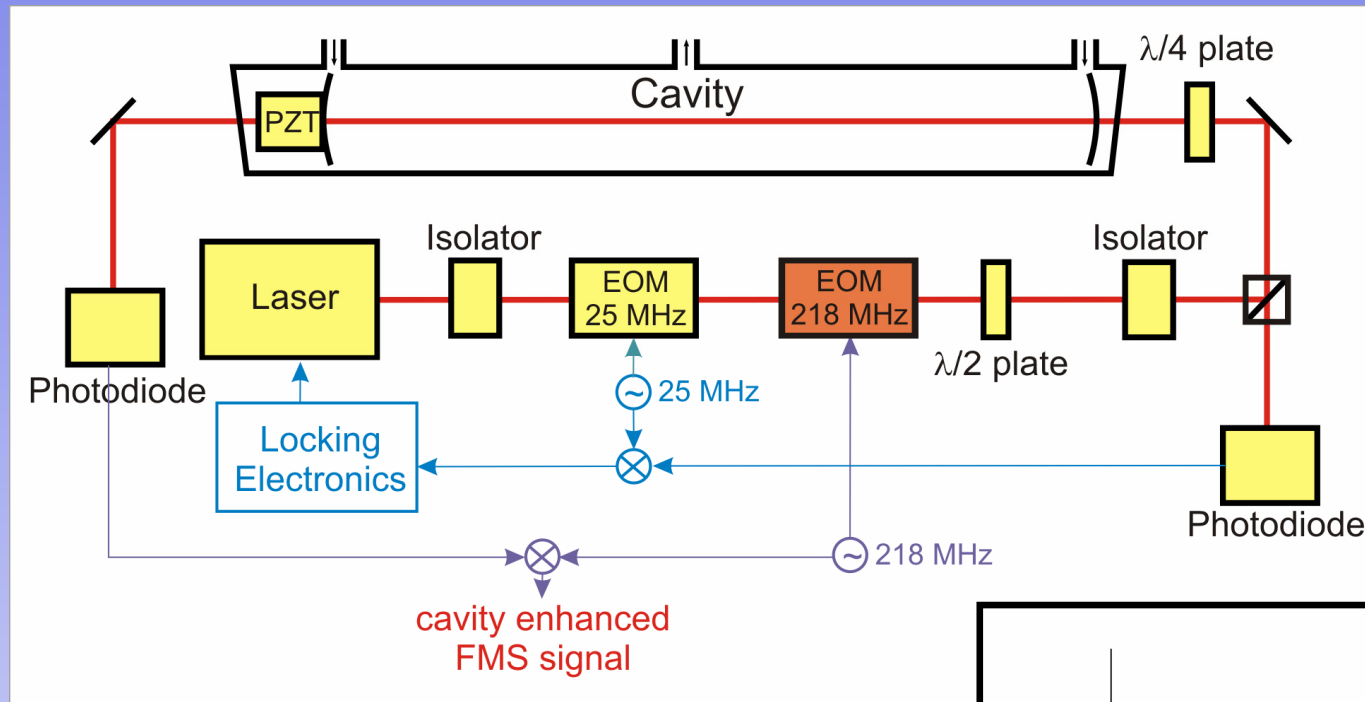


- Frequency modulation spectroscopy (FMS)
- Decreased noise with detection at high frequency

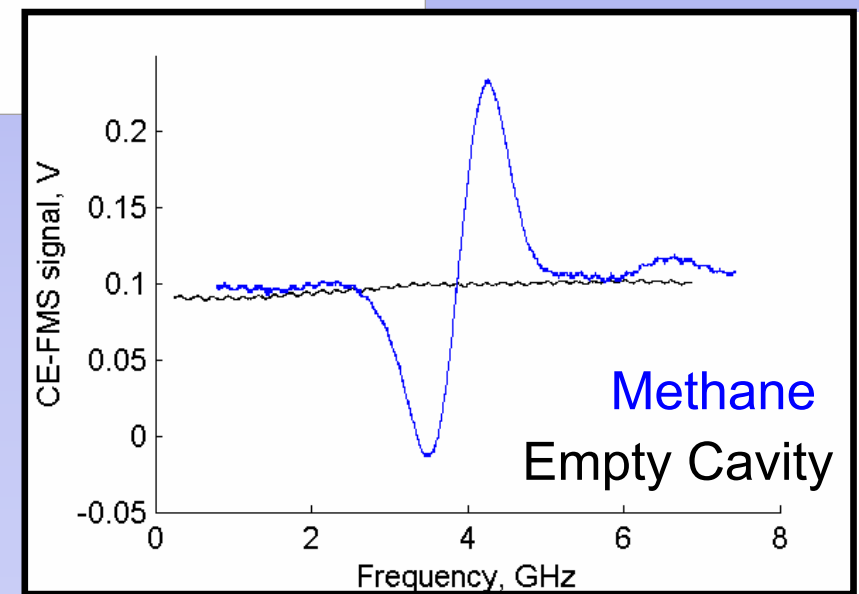


Oxford Experiment – Step 2

Cavity enhanced FMS

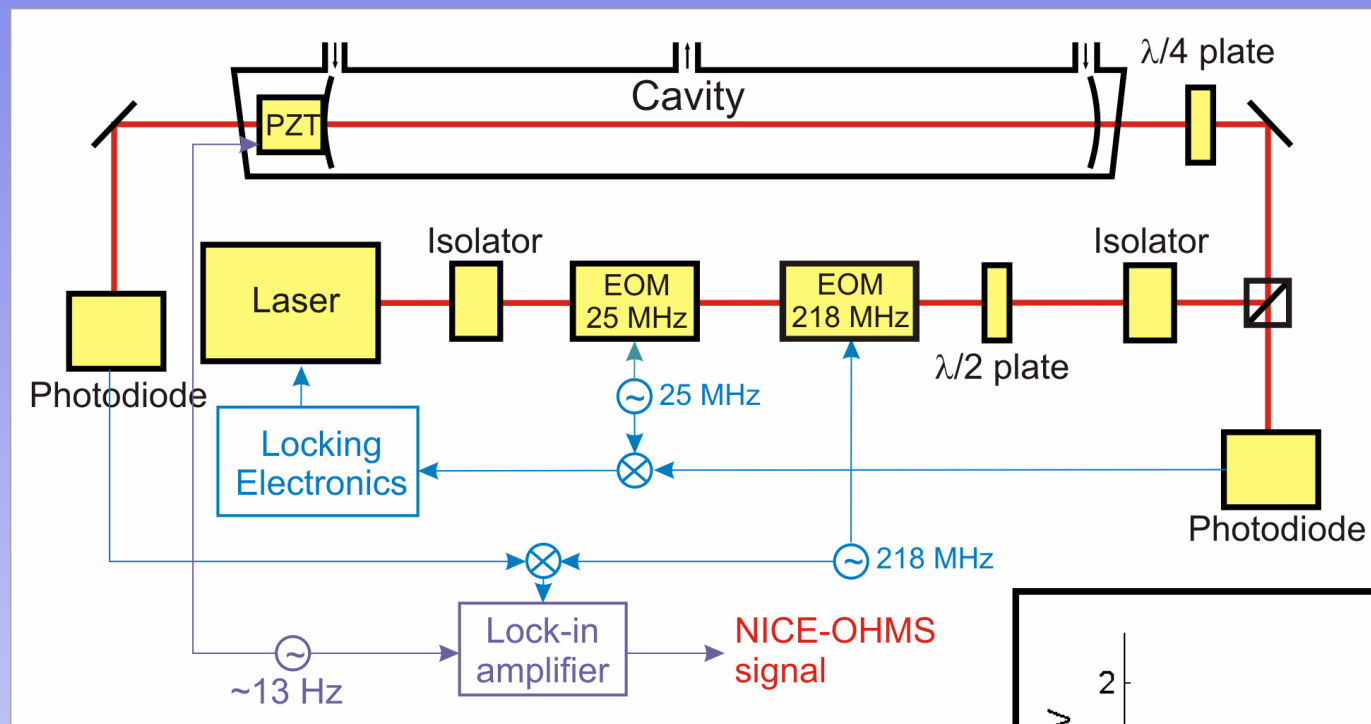


- FMS signal decoded in normal way for absorption profile
- Sensitivity of $5.13 \times 10^{-9} \text{ cm}^{-1}$

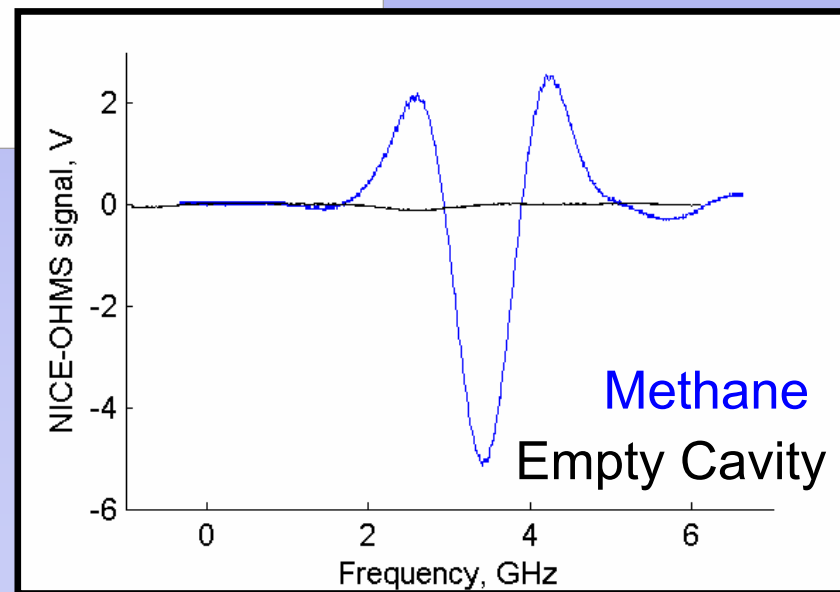


Oxford Experiment – Step 3

NICE-OHMS



- Extra layer of modulation removes noise source inherent to cavity enhanced FMS
- α_{\min} of $1.58 \times 10^{-9} \text{ cm}^{-1}$



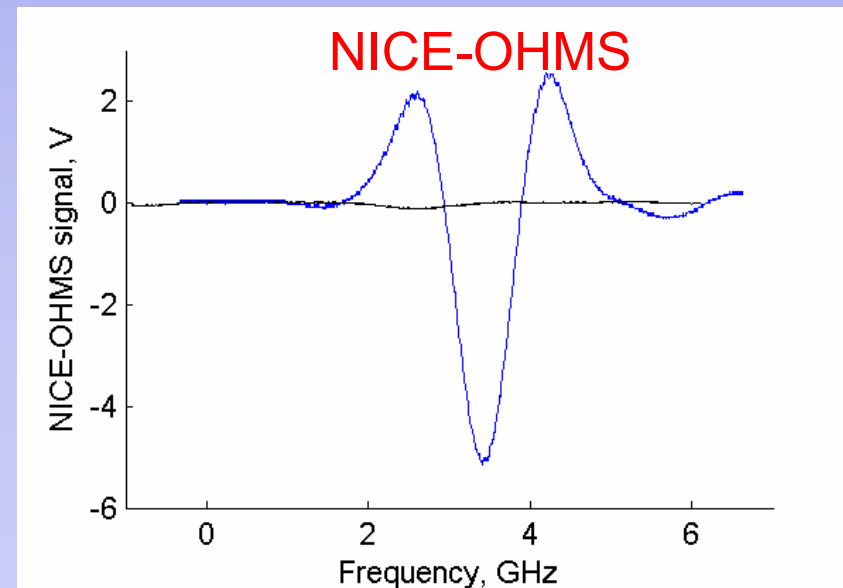
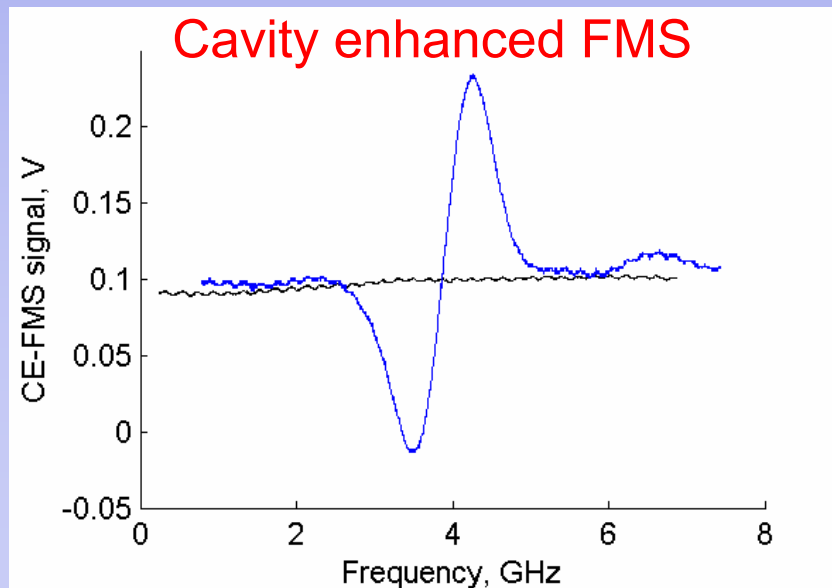
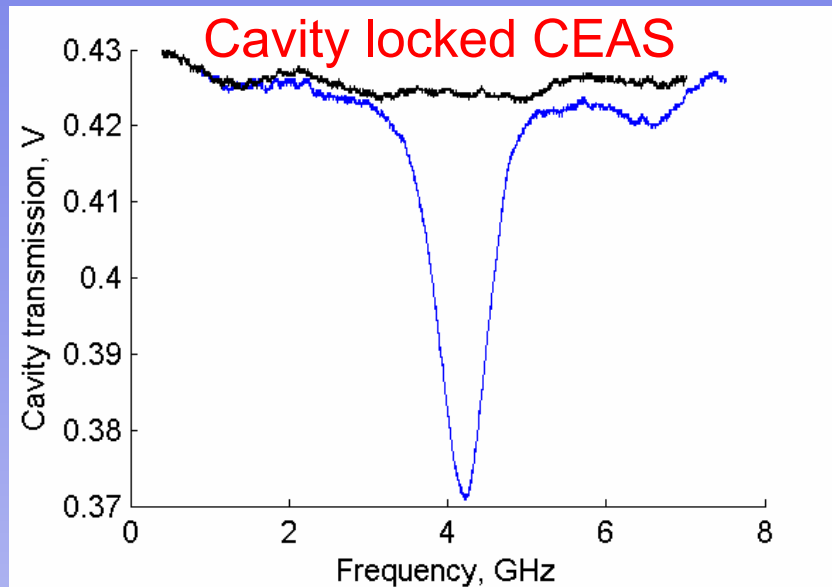
Cavity locked CEAS \rightarrow NICE-OHMS

Methane Detection

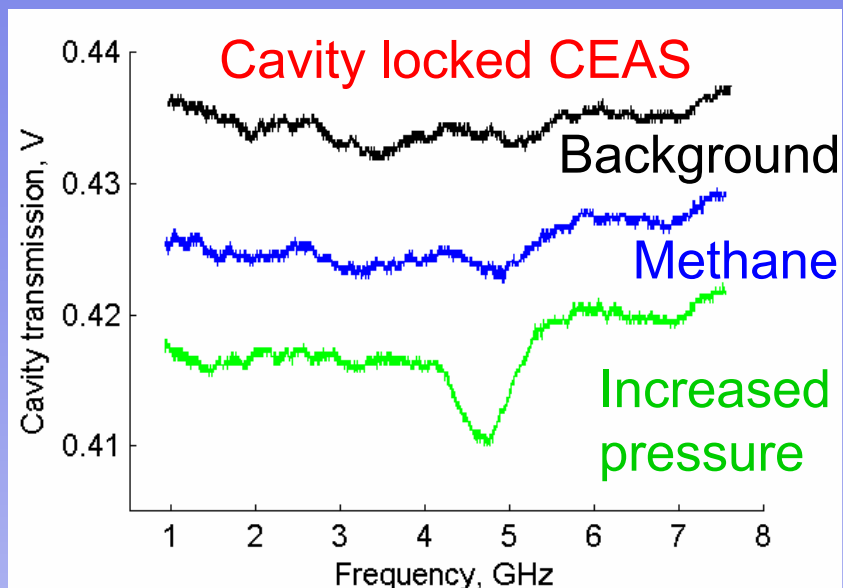
- 4.9 Torr
- unassigned transition at 6595.90 cm^{-1}
- Peak absorption of $7.7 \times 10^{-8} \text{ cm}^{-1}$

Minimum detectable absorption

- CL-CEAS $8.37 \times 10^{-9} \text{ cm}^{-1}$
- CE-FMS $5.13 \times 10^{-9} \text{ cm}^{-1}$
- NICE-OHMS $1.58 \times 10^{-9} \text{ cm}^{-1}$
($9.83 \times 10^{-10} \text{ cm}^{-1} \text{ Hz}^{-1/2}$)



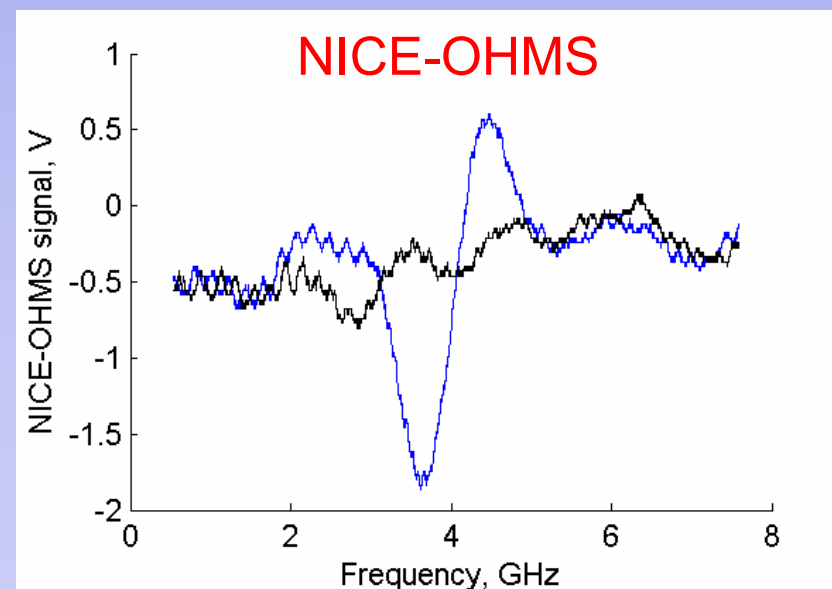
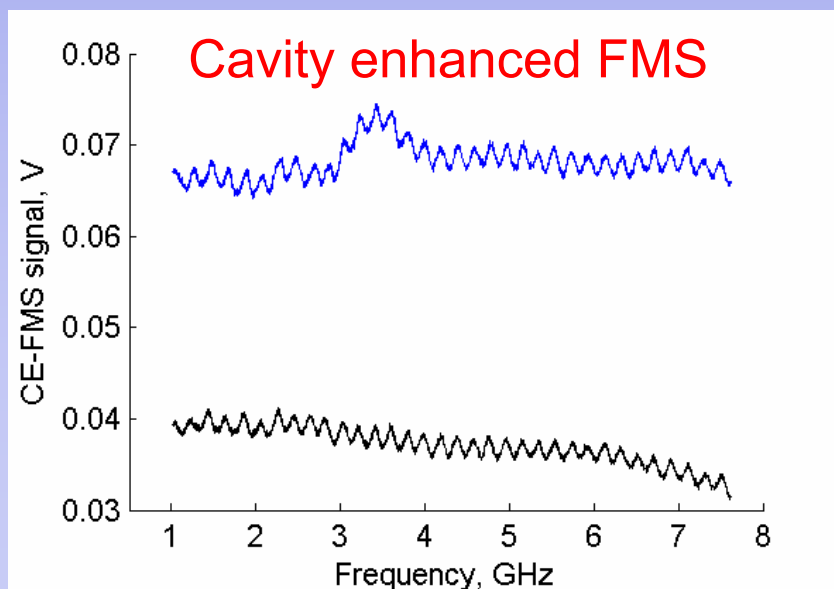
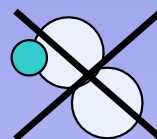
What is limiting the sensitivity?



- Reduce pressure to 160 mTorr
- Limiting noise sources become clear

Minimum detectable absorption

- NICE-OHMS $1.58 \times 10^{-9} \text{ cm}^{-1}$
 ~~$(9.83 \times 10^{-10} \text{ cm}^{-1} \text{ Hz}^{-1/2})$~~

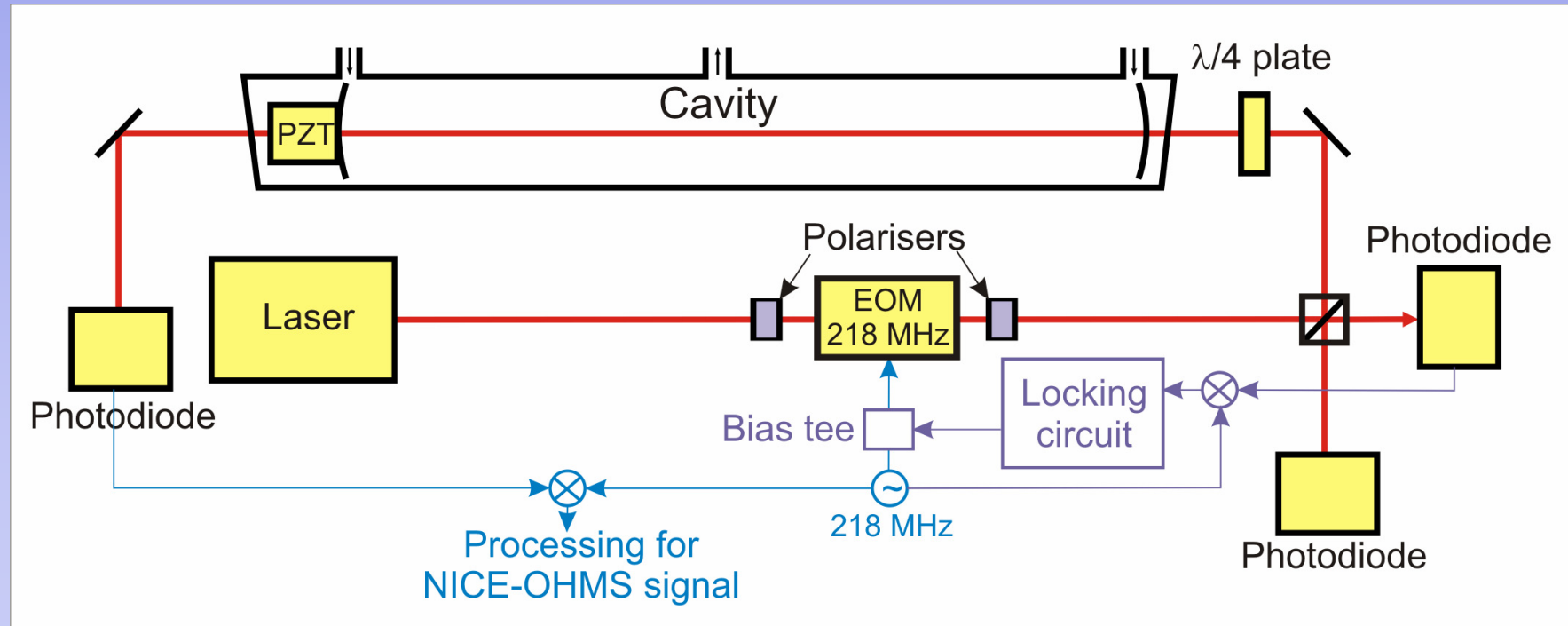


Planned Improvements 1

Removing RAM

Residual amplitude modulation (RAM)

- EOM provides phase modulation
- ... can provide amplitude modulation too!

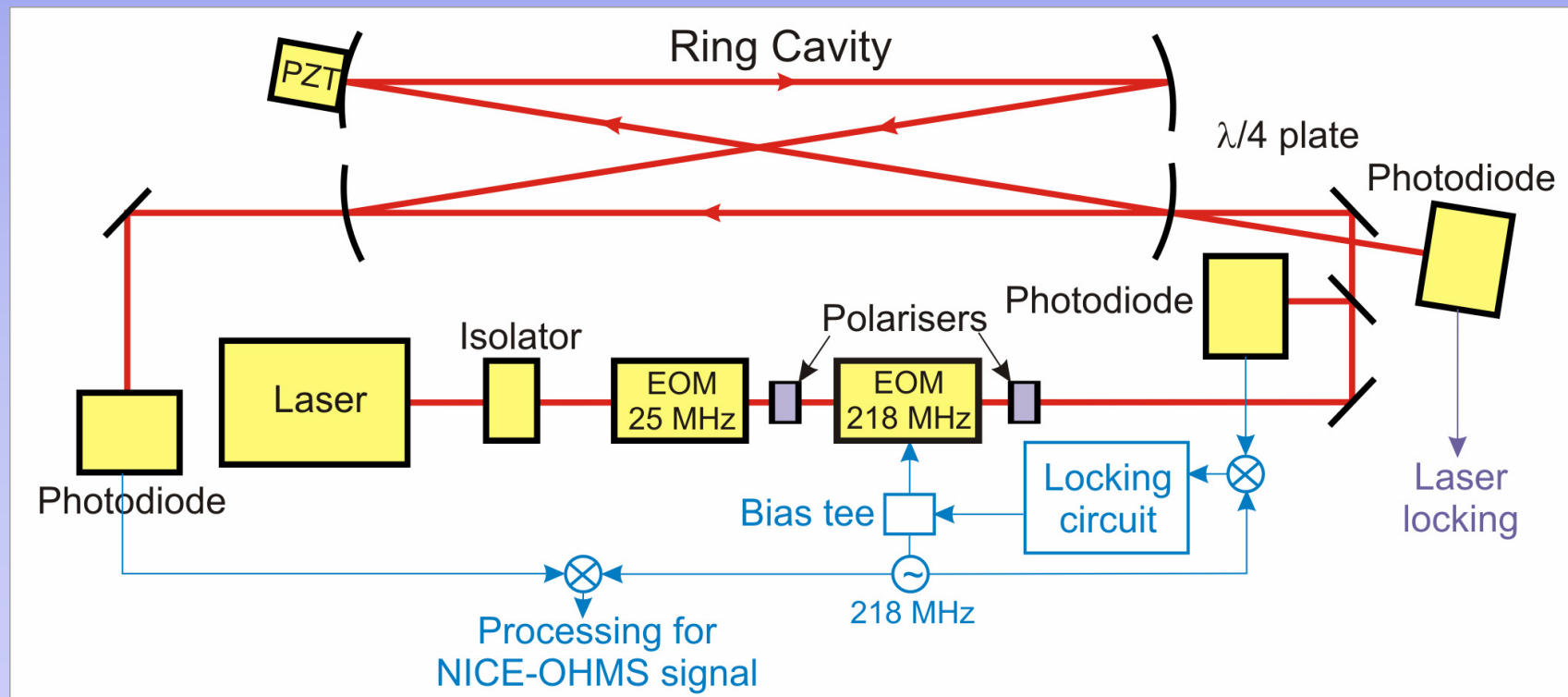


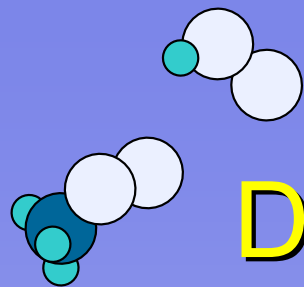
Planned Improvements 2

Ring Cavity

Etalons

- Replace linear cavity with ring cavity
- Stop formation of etalons involving cavity mirrors





Plans and Prospects: Detection of peroxy radicals



Plans

- Implement RAM removal locking and ring cavity
- Detect HO₂ at higher concentrations and investigate potential detection transitions

Prospects

- Potential sensitivity with ring cavity (Finesse 3300)?

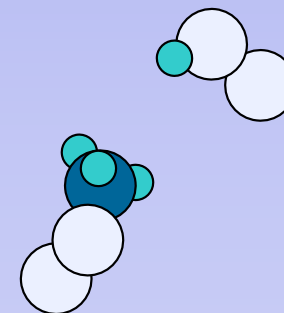
$$\alpha_{\text{shot noise}} = 4.3 \times 10^{-13} \text{ cm}^{-1} \text{ Hz}^{-1/2}$$

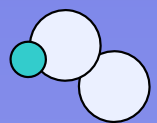
- Enough for detection of atmospheric HO₂?

- Yes! – with averaging or better mirrors

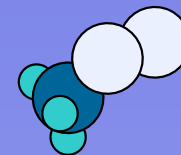
- Can we reach this ?

- We don't know – but we can try





Acknowledgements



Thanks to

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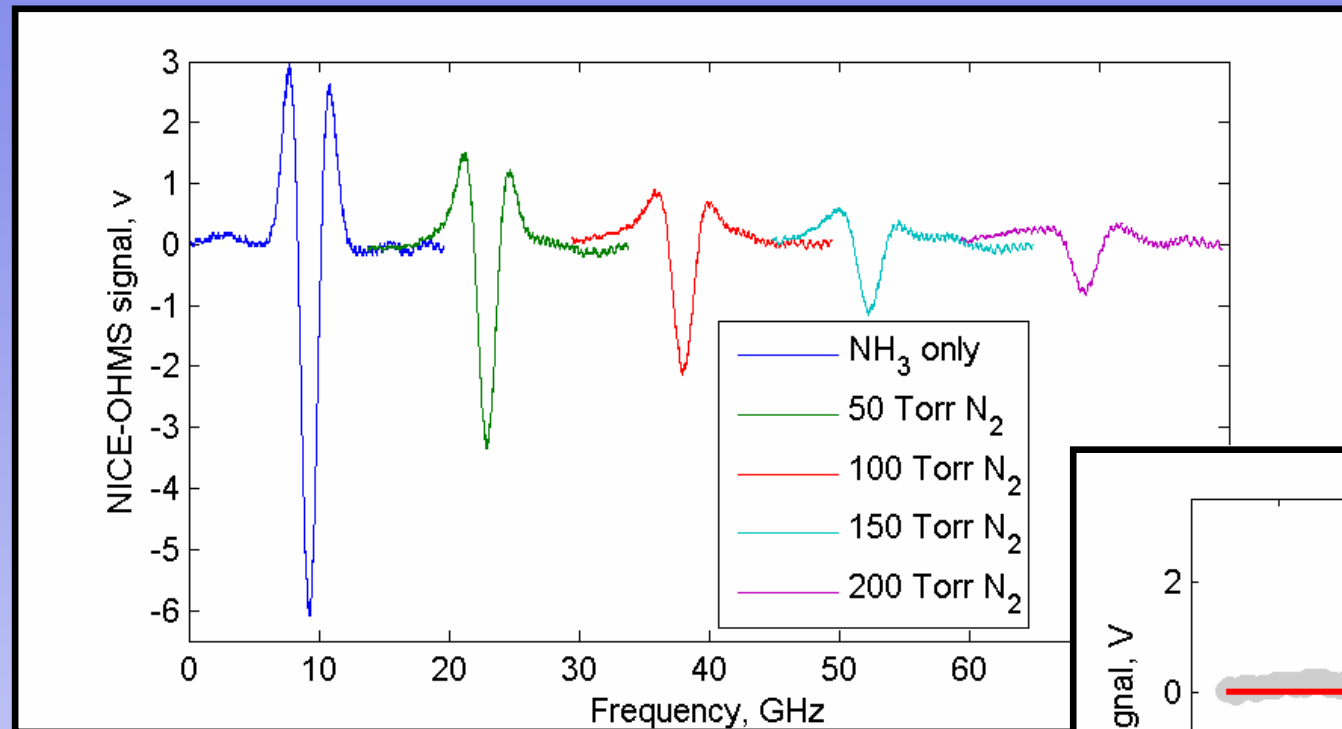
New Zealand Foundation
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and Technology

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Royal Society

EPSRC - Laser Portfolio
Award



NICE-OHMS signals: Broadening and Fitting



- Effect of pressure broadening on NICE-OHMS signal

- Use frequency and wavelength modulation theories to determine NICE-OHMS signal shape
- Fitting allows line parameters to be quantitatively determined

